Lead and Copper Research Update: Recent Findings and Future Areas of Work

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Acknowledgements

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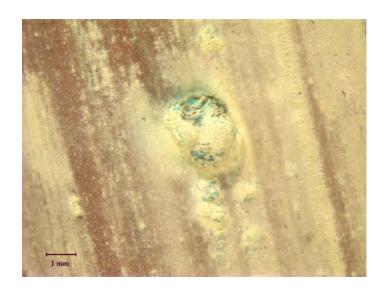
Copper and Lead Research Program

- Non-uniform copper corrosion (pitting)
 - Materials/solids analysis
 - Electrochemical corrosion studies
- Copper
 - Aging
 - High alkalinity
- Lead
 - Pb(II) vs Pb(IV)

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Non-Uniform Corrosion Copper Pitting Background

- Lead to "pinhole" leaks
- Costly Repairs
- Leaks may go undetected in walls or basements, and service lines
- Pinhole Leaks
 - Mold and Mildew
 - Liability Issues
- Does not lead to high copper levels at the tap



PITTING is a localized acceleration of corrosion that results in the thinning of the pipe wall in the impacted area.

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Pinhole Leaks

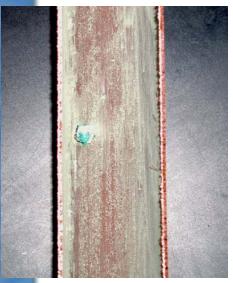


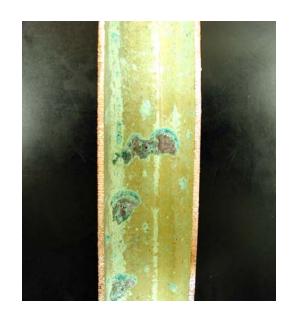


Pinhole leaks resulting from copper pitting

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Pitting Is Complicated



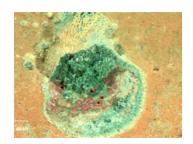




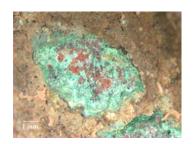
Ohio site #1



Wisconsin



Ohio site #2



All micrographs taken at 10x

decisions

Studying a Pitting Problem Approach

·Field Observations

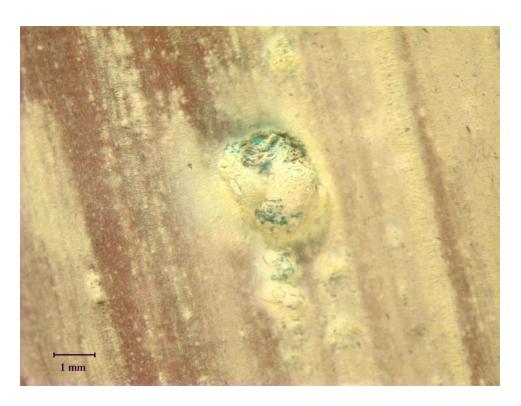
- Survey Form- plumbing type, cold vs. hot, horizontal, etc..
- ·Solids Analysis
 - ·SEM, EDS, XRD, etc..
- Water Quality
 - ·Hot vs. cold, similar waters

Case Study- Ohio Site Field Observations

- Cold water
- · Horizontal runs of pipe
- 3/4 pipe
- · Homes are about 7 years old
- Leaks occur near elbows and joints as well as in long runs
- No preference for the top or bottom of a pipe

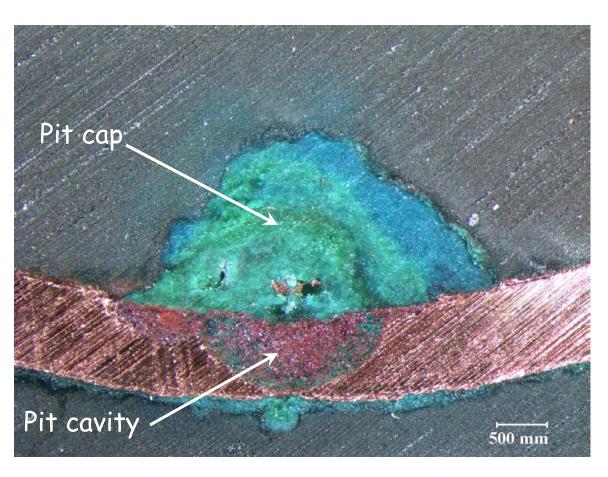
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Anatomy of a Pit Solids Analysis



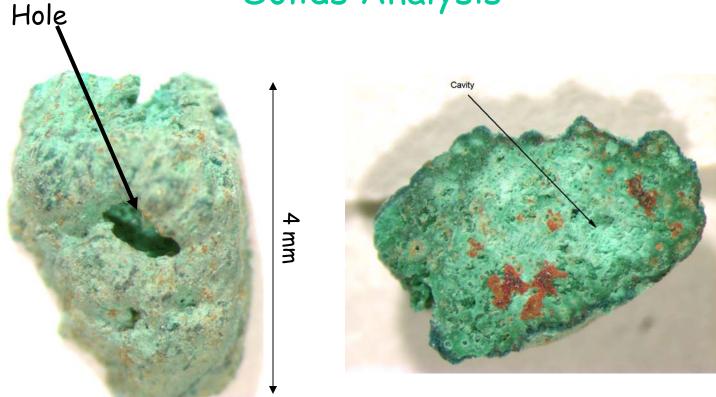
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Anatomy of a Pit Solids Analysis



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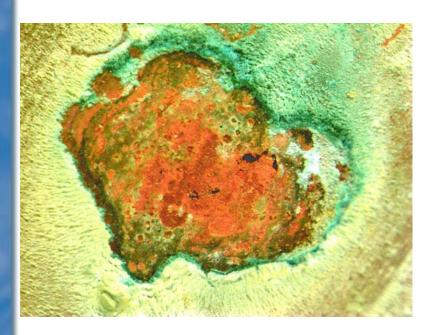
The Corrosion Cap Solids Analysis

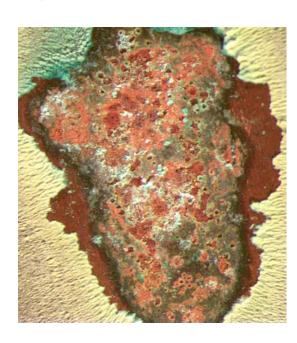


Brochantite - $Cu_4(OH)_6(SO_4)$ Ponsjakite - $Cu_4(OH)_6(SO_4)^*H_2O$

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Perforated Membrane Solids Analysis

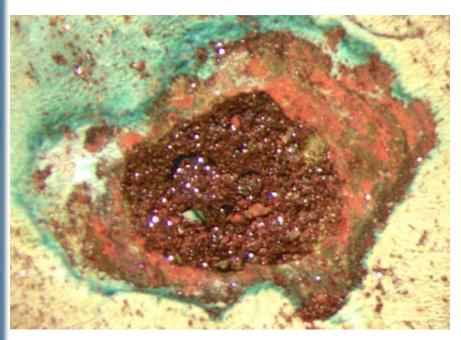


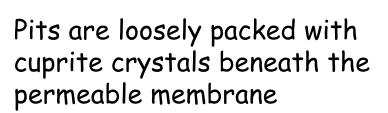


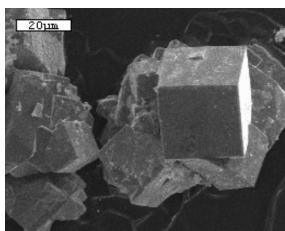
Literature suggests that the membrane consists of cuprite.

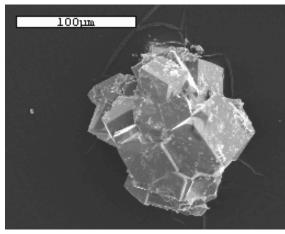
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Pit Cavity Solids Analysis









RESEARCH & DEVELOPMENT

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500 micron

decision

Pit Propagation Theory







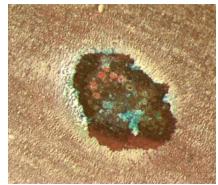


Particle deposition, particle growth, and corrosion cell formation



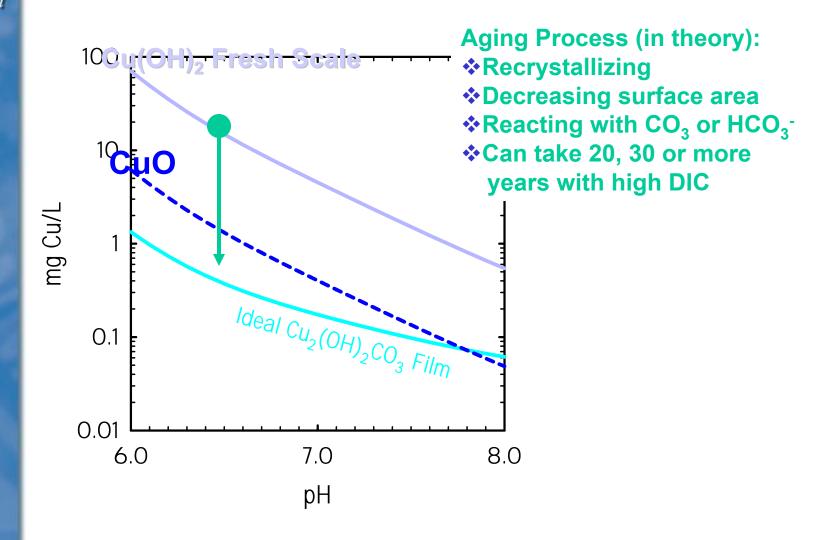






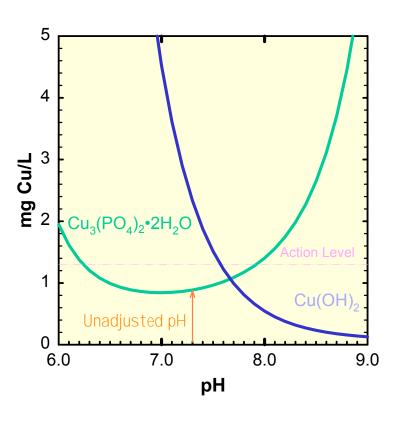
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Evolution of Scale Model for High DIC, Low pH



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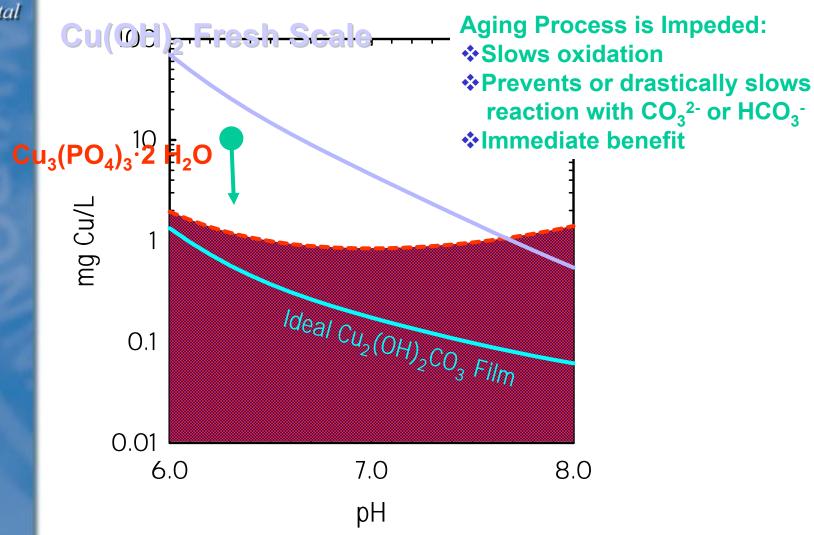
Theoretical Effect of Phosphate on Copper 3.3 mg PO₄/L at DIC=63 mg C/L





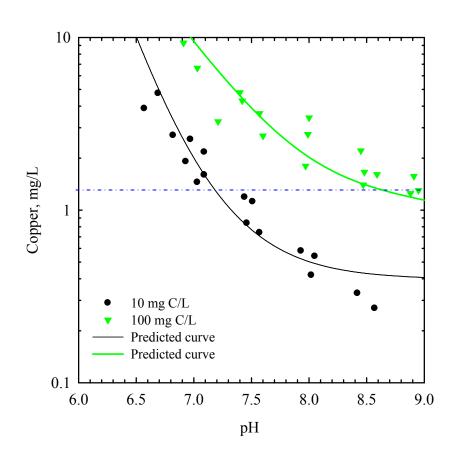
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Orthophosphate Effect on Scale Evolution at High DIC



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Effect of Dissolved Inorganic Carbon on Copper Solubility



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Pb Scales





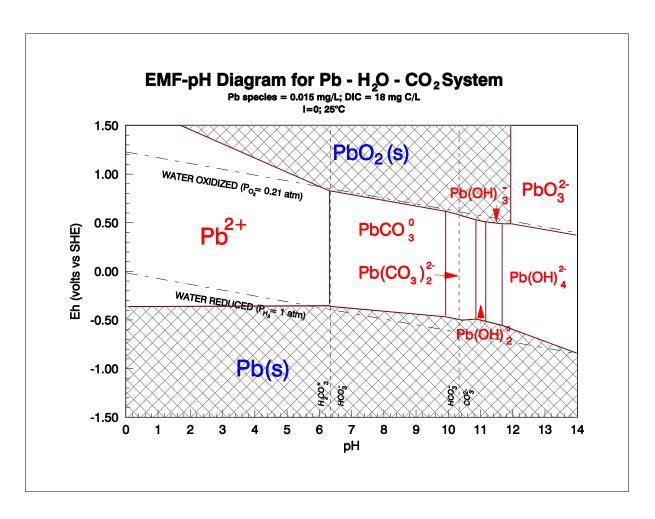


Lead Corrosion (Solubility)

- Current basis for lead control strategies is based on Pb(II) chemistry.
- Relatively recently, Pb(IV) solids (PbO_2) have been found in a several distribution systems.
- Pb(IV) chemistry in water is relatively unknown (battery industry most applicable).

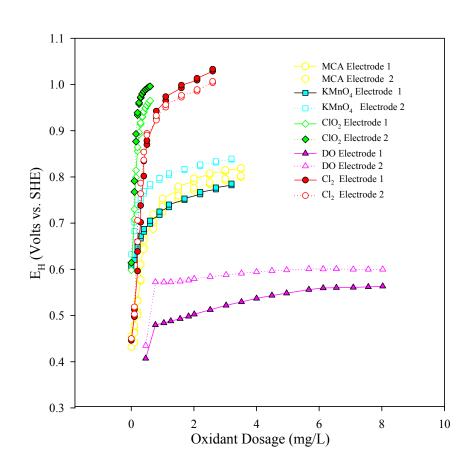
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Lead Redox Chemistry



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Redox Potential of Various Oxidants in Water at pH 7 10 mg C/L, 25°C



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Implications of Pb(IV)

- Pb(IV) can form in drinking water DS under high and persistent Eh
- Based on limited data Pb(IV) is much less soluble than Pb(II)
- How does Pb(IV) form?
- How sensitive is PbO₂ to redox potential?

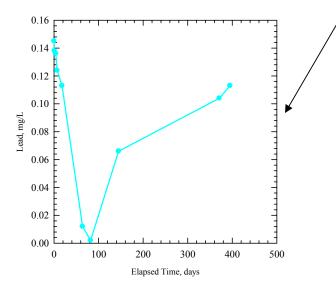
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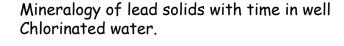
Bench Scale Studies

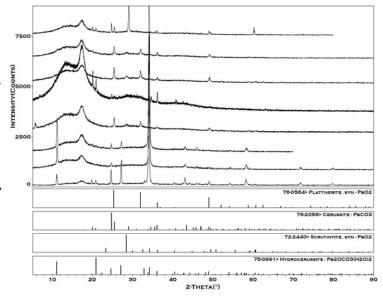
Lead Aging Study
What is the Role of ORP on Lead
Geochemistry in Water?



Appearance of lead solids with time







Corresponding change in soluble lead

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Conclusions

- Lead and copper continues to be an area of concern and research interest
- Copper pitting leading to pinhole leaks is a complicated and under-reported problem
- Uniform copper corrosion issues include: role of phosphate, high alkalinity (DIC)water and aging of minerals
- Lead(II) versus Pb(IV) is the latest lead issue

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Thank You